Synthesis of 3-(ω -Phenylalkyl)catechols, Phenolic Lipids found in Sap of the Burmese Lac Tree, via Directed Metallation

Yoshiaki Furukawa, Yoshiro Yamagiwa, and Tadao Kamikawa*

Department of Chemistry, Faculty of Science & Technology, Kinki University, Kowakae, Higashi-Osaka 577, Japan

3-(ω-Phenylalkyl)catechols were synthesized *via* directed metallation.

The phenolic lipids 3-(12-phenyldodecyl)catechol (1a) and 3-(10-phenyldecyl)catechol (1b) were recently isolated from the sap of the Burmese lac tree, *Melanorrhoea ustiate*, but neither their biological activities nor chemical properties are known. We tried to synthesize these lipids to carry out further studies on them. In an earlier work on synthesizing 3-alkyl-

catechols, e.g. urushiol, Yukawa et al.² alkylated catechol bistetrahydropyranyl ether via directed metallation but the yield was not very good. In this communication, we describe a new method of synthesizing these phenolic lipids via directed metallation using methoxymethyl ether as the directing group.³

HO(CH₂)_nOH
$$\stackrel{i,ii}{\longrightarrow}$$
 THPO(CH₂)_{n-1}CHO

(2) (3)
$$\downarrow^{iii}$$
HO(CH₂)_nPh $\stackrel{iv,v}{\longleftarrow}$ THPO(CH₂)_{n-1}CH(OH)Ph

(5) (4)
$$\downarrow^{ii}$$
OHC(CH₂)_{n-1}Ph

(6)
$$THP = \text{tetrahydropyranyl}$$
 $\mathbf{a}; n = 12, \mathbf{b}; n = 10$

Scheme 1. Reagents: i, dihydropyran, p-MeC₆H₄SO₃H, CH₂Cl₂; ii, (COCl)₂, Me₂SO, CH₂Cl₂, Et₃N; iii, PhMgBr, Et₂O; iv, Et₃SiH, CF₃CO₂H; v, NaOH, MeOH.

Monoprotection of dodecane-1,12-diol (2a) with dihydropyran and a catalytic amount of $p\text{-MeC}_6H_4SO_3H$ followed by oxidation by Swern's procedure⁴ gave the protected aldehyde (3a). Treatment of (3a) with phenylmagnesium bromide gave the alcohol (4a) [30% from (2a)]. Hydrogenolysis of (4a) with triethylsilane and trifluoroacetic acid⁵ followed by hydrolysis with alkali gave the alcohol (5a) (59%). Oxidation of (5a) by Swern's procedure⁴ gave the aldehyde (6a) (87%).

When the guaiacol methoxymethyl ether (7) was treated with n-butyl-lithium in diethyl ether at room temperature, a white precipitate formed and warming the mixture at 30 °C for 30 minutes resulted in an orange suspension. Addition of the aldehyde (6a) to this suspension gave the alcohol (8a) (49%). Ionic hydrogenation by Kursanov's procedure⁵ gave the phenol (9a) (30%). Treatment of (9a) with boron tribromide gave 3-(12-phenyldodecyl)catechol (1a) (84%), which had physical data identical with those reported by Oshima, thus confirming the structure of the natural product.

3-(10-Phenyldecyl)catechol (1b) was also prepared by a similar method from decane-1,10-diol (2b) in 2.3% overall yield.

Scheme 2. Reagents: i, BuⁿLi, Et₂O, 30 °C; ii, (6); iii, Et₃SiH, CF₃CO₂H; iv, BBr₃, CH₂Cl₂.

a; n = 12, b; n = 10

We thank Dr. Oshima for the spectral data of the natural products, and the Ministry of Education, Science and Culture, and Kinki University for the financial support of this work.

Received, 6th May 1986; Com. 597

References

- 1 Y. Du, R. Oshima, Y. Yamauchi, and J. Kumanotani, J. Chem. Soc., Chem. Commun., 1985, 630.
- 2 T. Hanafusa and Y. Yukawa, Nippon Kagaku Zasshi, 1966, 87, 202.
- 3 I. Kubo, T. Kamikawa, and I. Miura, Tetrahedron Lett., 1983, 24, 3825.
- 4 A. J. Mancuso, S.-L. Huang, and D. Swern, J. Org. Chem., 1978, 43, 2480.
- 5 D. N. Kursanov, Z. N. Parnes, and N. M. Loin, Synthesis, 1974, 633.